

L'accès en synthèse à ce type de structure hétérocyclique peu banale sera décrit dans une prochaine publication.

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Structure of (4*S*,5*S*,7*S*,8*S*)-11-Hydroxy-1(10)-guaïen-8-yl (1*S*)-1-Phenylethylcarbamate

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Abstract. (1) [3*S*-(3 α ,3 α ,5 α ,6 β)]-1,2,3,3a,4,5,6,7-Octahydro-3,8-dimethyl-5-(1-hydroxy-1-methylethylidene)azulen-6-yl (1*S*)-1-phenylethylcarbamate, C₂₄H₃₅NO₃, $M_r = 385.55$, monoclinic, $P2_1$, $a = 8.388$ (2), $b = 18.520$ (5), $c = 14.132$ (3) Å, $\beta = 98.71$ (2)°, $V = 2170$ (2) Å³, $Z = 4$, $D_x = 1.180$ (2) g cm⁻³, $\lambda(\text{Mo } K\alpha) = 0.71073$ Å, $\mu = 0.715$ cm⁻¹, $F(000) = 840$, $T = 110$ K, final $R = 0.051$ for 3360 unique observed reflections. The asymmetric unit contains molecules of approximately the same conformation, only the arrangement of the phenyl rings is slightly different. An intramolecular hydrogen bond is observed in both molecules, O(16)—O(19) = 2.80 (1) Å, O(16)—H(16A)—O(19) = 165 (1)° and O(16')—O(19') = 2.84 (1) Å, O(16')—H(16')—O(19') = 172 (1)°. The cycloheptene moiety adopts a twisted conformation which is a low energy conformer according to Allinger & Sprague [*J. Am. Chem. Soc.* (1972), **94**, 5734–5747]. The cyclopentane moiety adopts a slightly distorted envelope conformation. A search for CC(C)OC(=O)NC(C)C fragments in the Cambridge Structural Database (release of July, 1989), giving a total of 82 molecules, showed a planar arrangement of the urethane fragments [OC(=O)N], the arrangement being *trans-trans* in 75 molecules and *trans-cis* in 7 molecules. The *trans-trans* arrangement is also favoured in the present compound.

Experimental. (I) was prepared from (4*S*,5*S*,7*S*,8*S*)-1(10)-guaïen-8,11-diol and (S)-(-)-1-phenylethylisocyanate (Lemmich, Smitt, Jensen & Christensen (1991)). Crystallization from ether–petroleum ether gave colourless plates, m.p. 401–402 K. $[\alpha]_D^{20} = +11.2$ °. Crystal dimensions 0.1 × 0.3 × 0.4 mm. Preliminary film data indicated the space groups $P2_1$ or $P2_1/m$, with the reflections hkl : $h + l = 2n + 1$ systematically weak but observable. $P2_1$ was chosen since the crystals are chiral. Diffraction data were collected on an Enraf–Nonius CAD-4 diffractometer. The crystal was cooled to 110 ± 0.8 K in a stream of N₂ gas. Unit-cell dimensions were determined from a least-squares fit of 25 reflections, well distributed in reciprocal space, $15 < \theta < 21$ °. 7518 reflections were measured with $\omega/2\theta$ scans, $2\theta \leq 50.0$ °, hkl ranges: h : 0 → 9, k : -21 → 21, l : -16 → 16. Three reflections (10 $\bar{3}$, 032 and 020) were measured every 2.8 h to check for decay of the crystal, which decay was not observed. The intensity data were corrected for Lorentz–polarization effects but not for absorption. They were averaged to 6078 independent reflections [$R_{int}(I) = 0.034$], 3360 of which with $F_o^2 > 3.0\sigma(F_o^2)$ were considered observed. The structure was solved by direct methods using *SHELXS86* (Sheldrick, 1990) and least-squares refined on F with the *SDP* package (Enraf–Nonius, 1981). The H atoms were located and included in

Table 1. Fractional atomic coordinates and isotropic thermal parameters (Å²)

Anisotropically refined displacement parameters are given in the isotropic equivalent form defined as: $B_{eq} = (4/3)[a^2\beta_{11} + b^2\beta_{22} + c^2\beta_{33} + (2abc\cos\gamma)\beta_{12} + (2accos\alpha)\beta_{13} + (2bccos\alpha)\beta_{23}]$.

| | x | y | z | B_{eq}/B_{iso} |
|--------|-------------|------------|------------|------------------|
| C(1) | 0.4025 (6) | 0.1445 | 0.5904 (3) | 1.29 (9) |
| C(2) | 0.4113 (7) | 0.1085 (3) | 0.4948 (3) | 1.7 (1) |
| C(3) | 0.5212 (7) | 0.1588 (3) | 0.4472 (4) | 2.8 (1) |
| C(4) | 0.6375 (7) | 0.1905 (3) | 0.5316 (4) | 1.7 (1) |
| C(5) | 0.5174 (6) | 0.2075 (3) | 0.6024 (3) | 1.43 (9) |
| C(6) | 0.6024 (6) | 0.2208 (2) | 0.7058 (3) | 1.26 (9) |
| C(7) | 0.5043 (6) | 0.2630 (3) | 0.7683 (3) | 1.43 (9) |
| C(8) | 0.3305 (6) | 0.2342 (2) | 0.7688 (3) | 1.06 (9) |
| C(9) | 0.2974 (6) | 0.1544 (2) | 0.7480 (3) | 1.34 (9) |
| C(10) | 0.3107 (6) | 0.1204 (3) | 0.6514 (3) | 1.22 (9) |
| C(15) | 0.7672 (7) | 0.1377 (3) | 0.5705 (4) | 2.3 (1) |
| C(11) | 0.5957 (6) | 0.2727 (3) | 0.8720 (3) | 1.5 (1) |
| C(12) | 0.6103 (7) | 0.2032 (3) | 0.9305 (4) | 1.9 (1) |
| C(13) | 0.7604 (7) | 0.3057 (3) | 0.8735 (4) | 2.1 (1) |
| O(16) | 0.5030 (4) | 0.3213 (2) | 0.9238 (2) | 2.06 (8) |
| O(17) | 0.2178 (4) | 0.2742 (2) | 0.6981 (2) | 1.34 (7) |
| C(18) | 0.1763 (6) | 0.3406 (2) | 0.7251 (3) | 1.41 (9) |
| O(19) | 0.2238 (4) | 0.3663 (2) | 0.8045 (2) | 1.55 (7) |
| N(20) | 0.0795 (5) | 0.3744 (2) | 0.6540 (3) | 1.25 (7) |
| C(21) | 0.0198 (6) | 0.4477 (2) | 0.6667 (3) | 1.24 (9) |
| C(22) | 0.0595 (7) | 0.4952 (3) | 0.5863 (3) | 1.9 (1) |
| C(23) | -0.1582 (6) | 0.4493 (3) | 0.6718 (3) | 1.6 (1) |
| C(24) | -0.2692 (7) | 0.4134 (3) | 0.6044 (4) | 2.0 (1) |
| C(25) | -0.4329 (7) | 0.4178 (3) | 0.6101 (4) | 2.3 (1) |
| C(26) | -0.4864 (7) | 0.4570 (3) | 0.6801 (4) | 2.5 (1) |
| C(27) | -0.3799 (7) | 0.4935 (3) | 0.7475 (4) | 2.2 (1) |
| C(28) | -0.2162 (6) | 0.4898 (3) | 0.7425 (3) | 1.8 (1) |
| C(14) | 0.2055 (7) | 0.0547 (3) | 0.6333 (4) | 1.9 (1) |
| C(1') | 0.0211 (6) | 0.6686 (2) | 0.8889 (3) | 1.28 (9) |
| C(2') | -0.0260 (7) | 0.6335 (3) | 0.9780 (4) | 2.3 (1) |
| C(3') | -0.1446 (7) | 0.6847 (3) | 1.0113 (4) | 2.2 (1) |
| C(4') | -0.2242 (6) | 0.7247 (3) | 0.9214 (3) | 1.6 (1) |
| C(5') | -0.0793 (6) | 0.7370 (2) | 0.8679 (3) | 1.31 (9) |
| C(6') | -0.1295 (6) | 0.7543 (2) | 0.7611 (3) | 1.22 (9) |
| C(7') | 0.0014 (6) | 0.7924 (2) | 0.7153 (3) | 1.22 (9) |
| C(8') | 0.1681 (6) | 0.7574 (2) | 0.7320 (3) | 1.11 (9) |
| C(9') | 0.1782 (6) | 0.6749 (3) | 0.7488 (3) | 1.27 (9) |
| C(10') | 0.1276 (6) | 0.6424 (2) | 0.8384 (3) | 1.28 (9) |
| C(15') | -0.3557 (7) | 0.6773 (3) | 0.8643 (4) | 2.2 (1) |
| C(11') | -0.0566 (7) | 0.8059 (3) | 0.6063 (3) | 1.7 (1) |
| C(12') | -0.0709 (6) | 0.7376 (3) | 0.5479 (3) | 1.7 (1) |
| C(13') | -0.2137 (7) | 0.8483 (3) | 0.5902 (4) | 2.4 (1) |
| O(16') | 0.0625 (4) | 0.8507 (2) | 0.5682 (2) | 2.20 (8) |
| O(17') | 0.2660 (4) | 0.7889 (2) | 0.8171 (2) | 1.16 (7) |
| C(18') | 0.3223 (6) | 0.8560 (3) | 0.8019 (3) | 1.56 (9) |
| O(19') | 0.3009 (4) | 0.8870 (2) | 0.7254 (2) | 1.78 (8) |
| N(20') | 0.3960 (5) | 0.8848 (2) | 0.8872 (3) | 1.82 (8) |
| C(21') | 0.4577 (6) | 0.9589 (2) | 0.8867 (3) | 1.4 (1) |
| C(22') | 0.4233 (6) | 0.9948 (3) | 0.9792 (3) | 1.9 (1) |
| C(23') | 0.6350 (6) | 0.9625 (3) | 0.8762 (3) | 1.44 (9) |
| C(24') | 0.7531 (7) | 0.9315 (3) | 0.9434 (3) | 1.8 (1) |
| C(25') | 0.9159 (7) | 0.9361 (3) | 0.9314 (4) | 2.0 (1) |
| C(26') | 0.9620 (7) | 0.9725 (3) | 0.8556 (4) | 2.5 (1) |
| C(27') | 0.8440 (7) | 1.0050 (3) | 0.7896 (4) | 2.3 (1) |
| C(28') | 0.6817 (7) | 0.9990 (3) | 0.7988 (4) | 2.3 (1) |
| C(14') | 0.2209 (7) | 0.5736 (3) | 0.8684 (4) | 2.0 (1) |

Table 2. Selected bond distances (Å) and angles (°)

| | Unprimed molecule | Primed molecule |
|-------------------|-------------------|-----------------|
| C(1)—C(2) | 1.518 (6) | 1.523 (7) |
| C(1)—C(5) | 1.507 (6) | 1.524 (7) |
| C(1)—C(10) | 1.318 (7) | 1.318 (7) |
| C(2)—C(3) | 1.536 (8) | 1.501 (8) |
| C(3)—C(4) | 1.539 (7) | 1.533 (7) |
| C(4)—C(5) | 1.556 (8) | 1.543 (8) |
| C(4)—C(15) | 1.504 (7) | 1.539 (7) |
| C(5)—C(6) | 1.546 (6) | 1.538 (6) |
| C(6)—C(7) | 1.513 (7) | 1.528 (7) |
| C(7)—C(8) | 1.552 (7) | 1.527 (7) |
| C(7)—C(11) | 1.558 (6) | 1.565 (6) |
| C(8)—C(9) | 1.525 (6) | 1.547 (6) |
| C(8)—O(17) | 1.468 (5) | 1.470 (5) |
| C(9)—C(10) | 1.523 (7) | 1.521 (7) |
| C(10)—C(14) | 1.501 (7) | 1.520 (7) |
| C(11)—C(12) | 1.525 (7) | 1.505 (7) |
| C(11)—C(13) | 1.509 (8) | 1.520 (8) |
| C(11)—O(16) | 1.457 (6) | 1.462 (6) |
| O(17)—C(18) | 1.349 (6) | 1.358 (6) |
| C(18)—O(19) | 1.228 (5) | 1.213 (5) |
| C(18)—N(20) | 1.347 (6) | 1.377 (6) |
| N(20)—C(21) | 1.468 (6) | 1.466 (6) |
| C(21)—C(22) | 1.512 (7) | 1.532 (7) |
| C(21)—C(23) | 1.506 (8) | 1.519 (8) |
| C(23)—C(24) | 1.395 (7) | 1.389 (7) |
| C(23)—C(28) | 1.395 (7) | 1.391 (7) |
| C(24)—C(25) | 1.390 (8) | 1.405 (8) |
| C(25)—C(26) | 1.356 (8) | 1.370 (8) |
| C(26)—C(27) | 1.380 (7) | 1.391 (7) |
| C(27)—C(28) | 1.387 (8) | 1.391 (8) |
| C(2)—C(1)—C(5) | 108.9 (4) | 108.1 (4) |
| C(2)—C(1)—C(10) | 123.0 (3) | 124.9 (4) |
| C(5)—C(1)—C(10) | 128.0 (4) | 126.9 (4) |
| C(1)—C(2)—C(3) | 103.7 (4) | 104.9 (4) |
| C(2)—C(3)—C(4) | 104.2 (4) | 105.8 (4) |
| C(3)—C(4)—C(5) | 100.4 (4) | 101.6 (4) |
| C(3)—C(4)—C(15) | 111.9 (4) | 110.2 (4) |
| C(5)—C(4)—C(15) | 113.6 (4) | 112.5 (4) |
| C(1)—C(5)—C(4) | 103.7 (4) | 103.6 (4) |
| C(1)—C(5)—C(6) | 115.2 (4) | 115.2 (4) |
| C(4)—C(5)—C(6) | 112.9 (4) | 113.2 (4) |
| C(5)—C(6)—C(7) | 114.9 (4) | 113.9 (4) |
| C(6)—C(7)—C(8) | 114.9 (4) | 116.0 (4) |
| C(6)—C(7)—C(11) | 112.0 (4) | 110.7 (4) |
| C(8)—C(7)—C(11) | 111.3 (4) | 111.2 (4) |
| C(7)—C(8)—C(9) | 118.4 (4) | 118.1 (4) |
| C(7)—C(8)—O(17) | 109.4 (4) | 110.1 (4) |
| C(9)—C(8)—O(17) | 106.1 (3) | 104.8 (3) |
| C(8)—C(9)—C(10) | 122.6 (4) | 120.2 (4) |
| C(1)—C(10)—C(9) | 125.1 (4) | 127.0 (4) |
| C(1)—C(10)—C(14) | 123.4 (4) | 121.4 (4) |
| C(9)—C(10)—C(14) | 111.5 (4) | 111.5 (4) |
| C(7)—C(11)—C(12) | 113.6 (4) | 113.1 (4) |
| C(7)—C(11)—C(13) | 112.3 (4) | 111.3 (4) |
| C(7)—C(11)—O(16) | 108.7 (4) | 109.1 (4) |
| C(12)—C(11)—C(13) | 109.7 (4) | 110.9 (4) |
| C(12)—C(11)—O(16) | 104.7 (4) | 106.0 (4) |
| C(13)—C(11)—O(16) | 107.4 (4) | 106.2 (4) |
| C(8)—O(17)—C(18) | 115.7 (3) | 113.4 (3) |
| O(17)—C(18)—O(19) | 123.5 (4) | 124.5 (4) |
| O(17)—C(18)—N(20) | 111.4 (4) | 109.7 (4) |
| O(19)—C(18)—N(20) | 125.2 (4) | 125.6 (4) |
| C(18)—N(20)—C(21) | 121.2 (4) | 118.1 (4) |
| N(20)—C(21)—C(22) | 109.2 (4) | 107.0 (4) |
| N(20)—C(21)—C(23) | 112.5 (4) | 113.2 (4) |
| C(22)—C(21)—C(23) | 111.0 (4) | 112.2 (4) |
| C(21)—C(23)—C(24) | 121.4 (5) | 121.4 (4) |
| C(21)—C(23)—C(28) | 120.1 (4) | 119.7 (4) |
| C(24)—C(23)—C(28) | 118.4 (5) | 118.9 (5) |
| C(23)—C(24)—C(25) | 119.7 (5) | 119.8 (5) |
| C(24)—C(25)—C(26) | 120.8 (5) | 121.3 (5) |
| C(25)—C(26)—C(27) | 121.0 (6) | 118.7 (5) |
| C(26)—C(27)—C(28) | 118.9 (5) | 120.6 (5) |
| C(23)—C(28)—C(27) | 121.2 (4) | 120.5 (5) |

Fourier summations. 255 parameters were refined to $R = 0.051$, $wR = 0.049$ and $S = 1.24$ with $w = [\sigma^2(F) + 0.005F^2]^{-1}$; max. and min. heights in final $\Delta\rho$ map are 0.59 (3) and -0.45 (3) e Å⁻³, $(\Delta/\sigma)_{\max} = 0.003$. Atomic scattering factors were taken from *International Tables for Crystallography* (1974, Vol. IV, Table 2.2B), except for H atoms for which they were taken from Stewart, Davidson & Simpson (1965). The number of observed reflections allowed for anisotropic refinement of the oxygen but not the other atoms.

The asymmetric unit contains two molecules of the title compound. In one of the molecules two positions for the hydroxylic H(16) atom are located in

the difference Fourier map. Both of these positions were included with a population parameter of 0.5. The two positions of H(16) can be explained by the possibility of a hydrogen bond being formed either intramolecularly to O(19) or intermolecularly to

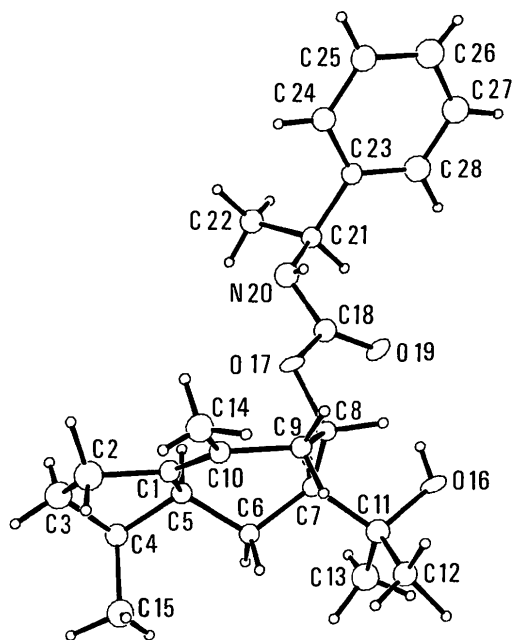


Fig. 1. ORTEP (Johnson, 1976) drawing of (4*S*,5*S*,7*S*,8*S*)-11-hydroxy-1(10-guaien-8-yl)-(1*S*)-1-phenylethylcarbamate. The thermal ellipsoids for non-H atoms are drawn at 50% probability level.

N(20'). Final atomic coordinates and isotropic or equivalent isotropic displacement parameters of non-H atoms are listed in Table 1.* Bond distances

* Lists of structure factors, anisotropic thermal parameters, bond distances and angles involving H atoms and H-atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 53861 (41 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Structure of a Model for Aspirochlorine (Antibiotic A30641)

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Abstract. (1) $C_{20}H_{18}N_2O_5S_2$, $M_r = 430.48$, monoclinic, $P2_1/c$, $a = 9.934$ (3), $b = 18.730$ (5), $c = 10.563$ (2) Å, $\beta = 96.23$ (2)°, $V = 1954$ (1) Å³, $Z = 4$, $D_x = 1.46$ g cm⁻³, $\lambda(\text{Mo } K\alpha) = 0.7107$ Å, $\mu = 3.0$ cm⁻¹, $F(000) = 896$, $T = 112$ K, $R = 0.046$ ($wR = 0.041$) for 2813 unique, observed reflections. The

and angles are given in Table 2. A perspective drawing of one of the molecules is shown in Fig. 1 giving the atomic numbering scheme.

Related literature. In a chemotaxonomic study of the genus *Thapsia* the distribution of sesquiterpenes and sesquiterpene lactones was investigated (Smitt, 1991). Within the species *Thapsia villosa* var. minor morphological as well as phytochemical differences were revealed. The occurrence of three new compounds, which are monoesters of a sesquiterpene diol with a guaian skeleton, in only some specimens is of chemotaxonomic value (Lemmich, Smitt, Jensen & Christensen, 1991). The structure of the three compounds was elucidated by spectroscopic methods, but it was not possible to establish the relative configuration at all the chiral centres in the sesquiterpene moiety from the NMR spectroscopic data.

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compound is a model for aspirochlorine (antibiotic A30641).

